



Metallic Abrasion

Make-up present on the skin or clothing is the most common cause of the blackening or smudging. Cosmetics often contain chemical compounds which are harder than the jewelry itself. Metallic abrasion occurs when these hard compounds come in contact with the jewelry metals and wear or rub off very tiny particles of metal which appear as jet black dust. Very finely divided metals always appear black rather than metallic. When this dust falls on absorbent surfaces, such as the skin or clothing it sticks, forming a black smudge.

To prevent this, cosmetics which contain zinc oxide, titanium dioxide, ferric oxide and calamine should be avoided. If this is not possible, we recommend that you remove rings and other metal jewelry while applying cosmetics. Also, clean with soap and water those skin areas which will be in contact with your jewelry.

Corrosion

Another cause of discoloration is due to the corrosion of the metals from which the jewelry is made. Pure gold itself will not corrode but is too soft for jewelry and therefore is alloyed with proportions of base metals - usually silver or copper.

The composition of 14K gold is 14/24 pure gold and 10/24 alloys. Only the base metal used for alloying corrodes, leaving the gold untouched. The chemical compounds of silver and copper which are formed turn very dark even when slightly corroded. This corrosion occurs under moist or wet conditions.

Our skin on which the jewelry lies, is provided with thousands of perspiration glands. Chemically the perspiration is mostly fat and fatty acids. These essentially mild chemicals are enough to cause corrosion of 14K gold especially when aided by warmth and free access of air. When this happens, we suggest that you remove your jewelry often and use an absorbent powder, free of zinc oxide, on the skin areas which come into contact with your jewelry.

Rings containing copper and silver alloys, particularly sterling silver, corrode readily on the skin if enough salt is present. Simply exposing the hands to salted nuts, crackers or popcorn can cause such corrosion and discolor the skin. Rings should be taken off when you must expose your hands to salt.

Tarnishing of Sterling Silver Jewelry

Sterling silver is a copper-silver alloy and is somewhat softer than most gold alloys. Therefore, it is more likely to be abraded to black dust by relative movement between the jewelry and the skin or clothing.

Sterling silver jewelry has a pronounced tendency to tarnish. The tarnish is always due to the contact with sulphur compounds. The main cause of tarnishing of silver is a trace of hydrogen sulfide gas in the air. This gas is particularly present in smoke from burnt raw fuel, such as coal or oil; in some strong smelling foodstuffs, such as eggs and onions; in some fish or shellfish; and in polluted air.

The tarnish may turn the jewelry black and may also be rubbed off on the skin or clothing. A few people with especially moist skin may find that the sterling silver stains their skin green. This is due to the copper component of the silver alloy. When this occurs you should polish your jewelry frequently and use an absorbent powder on those moist skin areas which come into contact with your jewelry.

Other Problems and Solutions

Soap, detergent, wax, polish and cosmetic creams may build up under a ring, causing a type of dermatitis. A burning sensation and redness of the skin results. To prevent this, you should remove all rings before using soap or detergent, and clean the rings frequently.

The wearing of silver jewelry when manipulating photographic chemicals or solutions may cause staining of the skin. When silver is attacked by a chemical solution such as nitric acid, the solution of silver salt formed is reduced back to silver by the skin. This silver is in a finely divided form and causes an intensely black stain. Silver jewelry should be removed before working with photographic chemicals or solutions.

We offer this information not as an excuse, but to clarify the reasons, provide the facts and offer solutions for those who may encounter the problems described.

Fisher, Dr. Alexander A., "What Causes 'Gold Smudge'"
Jewelers Circular-Keystone, March 1971

Gardam, D.E., "Why Jewellery Sometimes Blackens the Skin or the Clothing", The Worshipful Company of Goldsmiths, Special Report No. 4, October 1969

The Look Of Platinum

The Cost Of Gold

For \$275 you get one platinum engagement ring.



For \$58 less, you get one engagement set PLUS one man's wedding band in TruPd™ 950 Palladium.*



NEW
TruPd™

*Price reflects any mounting (without diamonds) of equal weight to platinum in the following markets.
Platinum - \$1000/Oz • Palladium - \$325/Oz • \$125/Oz